

CROCOPY RESOLUTION TEST CHART
MITTONAL BUREAU OF STANDARDS-1963-A

FILE COPY

OTIC

Final Scientific Report

Submitted to:

Department of the Air Force AFOSR/NC Contract No. AFOSR-83-0291

Authors:

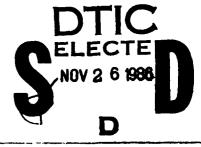
H.A. McKinstry, D.K. Agrawal,G.E. Lenain, C.S. Vikram,S.Y. Limaye, and A. Patankar

Approved for public release; distribution unlimited.

MAINTON OFFICE OF SCIENTISTO ECSEARCH (AFE

Approved for public released
Distribution Unlimited

August 1986





### MATERIALS RESEARCH LABORATORY

THE PENNSYLVANIA STATE UNIVERSITY UNIVERSITY PARK, PENNSYLVANIA 16802

86 11 25 286

٠					
ı	ECHBITY	CI ASSIFIC	ATION	OF THIS	PAGE

	LASSIFICATI	ON OF THIS PAGE					
REPORT DOCUMENTATION PAGE							
1a REPORT SECURITY CLASSIFICATION				16. RESTRICTIVE M	ARKINGS		
2s. SECURI	TY CLASSIFI	CATION AUTHORITY		3. DISTRIBUTION/AVAILABILITY OF REPORT			
				Approved for public release,			
26. DECLA	BIFICATION	DOWNGRADING SCHEE	OULE	distribution unlimited  5. MONITORING ORGANIZATION REPORT NUMBER(S)			
4. PERFOR	MING ORGAN	IZATION REPORT NUM	SER(S)				
				AFOSR-	TR. RA	-2060	
		NG ORGANIZATION	Bb. OFFICE SYMBOL (If applicable)	74. NAME OF MONITORING ORGANIZATION			
Univer	nnsylvani	la State	(ii) application	AROCD		•	
		and ZIP Code)	<u> </u>	AFOSR  7b. ADDRESS (City, State and ZIP Code)			
		Research Laborat	orv	Bldg. 410			
	•	c, PA 16802		Bolling AFB, DC 20332			
02001.	Jac, 100.	., 20002		bolling in	D, DO 2000	_	
	F FUNDING/	SPONSORING	Sb. OFFICE SYMBOL	9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER			
Dept.	of the Ai	lr Force	NC	AFOSE	1-83-0291		
		and ZIP Code)		10. SOURCE OF FUR	NDING NOS.		
		e of Scientific	Research	PROGRAM	PROJECT	TASK	WORK UNIT
)	NC Bldg.			ELEMENT NO.	NO.	NO.	NO.
	AFB, DO	y Classification)		61102p	2202		1
1		<u>al Expansion Ce</u>	ramics	61102F	2303	A3	
	AL AUTHOR		Lamics	·	<u></u>	<del></del>	<u></u>
H.A. Mo	Kinstry.	D.K. Agrawal.	G.E. Lenain. C.S	S. Vikram, S.Y	. Limaye.	A. Patankar	
Final I	F REPORT	135 TIME C	OVERED	14. DATE OF REPO!	AT (Yr., Mo., Day)	15. PAGE C	OUNT
	MENTARY N		TO	1.08050 1300			
10. 007766		J'A'ION					
17.	COSATI	CODES	18. SUBJECT TERMS (C				
FIELD	GROUP	SUB. GR.	Low Thermal Exp		• •	-	
		<u></u>	X-ray Diffract:	ion Thermal Ex	tpansion, T	hermal Expan	nsion of
10 400704	CT (Continue		Perovskites.				44
19. 200	C (Community	On reperse If necessary and	identify by block number	M Supersci	ut I ) W	SK DEL SCLUP	11/
6	ne cry	stal chemistry,	synthesis, and	Thermal expan	sion inves	tigation of	(MZ)
ram:	Co Sr	ompletedin th	at Tr2P3012 to	r = Ll, Na, K	Single	ervetale of	(NZD)
			igh-temperature				
crystals to test the validity of a structural model to interpret thermal expansion data. The possibility of the development of a glass-ceramic of NaGe2P3012 was ex-							
plored. Dielectric measurements made on CaZr <sub>4</sub> P <sub>6</sub> O <sub>24</sub> reveal that in general [NZP]-							
materials have a low dielectric constant.							
Or The							
	Three new families, namely diborides (ZrB2, TiB2, and CrB2), A1203-GeO2 system and perovskite $Pb(Mg_{1/3}Nb_{2/3})0_3$ were investigated in order to search for any new low						ystem
and	perovski	te Pb (Mg1/3Nb2/	3)03 were invest	igated in ord	er to sear	ch for any n	ew low
ther	thermal expansion composition, but not much success was achieved in this direction, except that some perovskite compositions displayed low a behavior at low temperatures.						
							ratures
20. DISTRIBUTION/AVAILABILITY OF ABSTRACT			21. ABSTRACT SECL	JRITY CLASSIFI	CATION		
UNCLASSIFIED/UNLIMITED - SAME AS RPT DTIC USERS -							
22s. NAME (	OF RESPONS	BLE INDIVIDUAL		22b. TELEPHONE NO		22c. OFFICE SYM	<b>8</b> 0L
	Dr. Ulrich				63	NC	

New magnetic materials with low thermal expansion characteristics were developed by (i) making ionic substitution in [NZP]-structure and (ii) by adopting a diphasic composite approach. Such a material was produced using Na<sub>4</sub>Zr<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> and Y<sub>3</sub>Fe<sub>5</sub>O<sub>12</sub>.

A di/multiphasic approach was adopted to tailor and control the thermal expansion of a ceramic, in this approach [NZP] was mixed with a second phase which is thermodynamically compatible. The sample was fabricated by a suitable heat treatment. Several compositions which demonstrate very low  $\alpha$  over a wide temperature range were produced.

Finally, several new techniques for measuring ultra-low thermal expansion with high precision and accuracy utilizing laser spectroscopy were developed.



Accesion	For	I			
NTIS C DTIC 1 Unanno Justifica	rAB unded	0			
L	By Dict ib: tion/				
A	Availability Codes				
Dist	Avail a Spe				
A-1	-				

#### TABLE OF CONTENTS

		rage
Summary.		.ii
ı.	Synthesis, crystal chemistry, and thermal expansion of [NZP]-family of materials	.ii
II.	Exploration for new families of low-expansion materials	.ii
III	. New low thermal expansion magnetic materials	.ii
IV.	Development of di/multiphasic micro- mposite ceramics for low $\alpha$ applications	
v.	New methods for measuring ultra-low thermal expansion.	iii
Final Re	port	. 1
ı.	Synthesis, Crystal Chemistry, and Thermal Expansion of [NZP]-Family	. 1
II.	Exploration of New Families of Low Expansion Materials	. 3
III	. New Low Thermal Expansion Magnetic Materials	. 3
IV.	Development of Di/Multiphasic Micro-composite Ceramics for Low $\alpha$ Applications	
v.	New Methods for Measuring Ultra-low Thermal Expansion.	. 4
Reference	es	. 5
Idst of I	Publications	. 7

#### SUMMARY

- I. Synthesis, crystal chemistry, and thermal expansion of [NZP]-family of materials: The work on [NZP] or [CTP] family can be subdivided into the following areas:
  - 1. Synthesis and thermal expansion of  $M^1A_2P_3O_{12}$  ( $M^1$  = Li, Na, K, Rb, Cs) and  $M^{11}A_4P_6O_{24}$  ( $M^{11}$  = Mg, Ca, Sr, Ba) where A stands for Zr and Ti.
  - 2. Single crystal growth of  $CaZr_4P_6O_{24}$  and  $NaZr_2P_3O_{12}$  by flux method.
  - Development of a glass-ceramic with [NZP]-composition.
  - 4. Dielectric studies of CaZr<sub>4</sub>P<sub>6</sub>0<sub>24</sub>.
  - 5. Investigation of crystalline constraints on thermal expansion of  ${\tt NaZr_2P_3O_{12}}.$
  - Exploratory substitution in the [NZP]-structure to find any new low α
    compounds.
- II. Exploration for new families of low-expansion materials: The following new families for possible low  $\alpha$  characteristics were investigated:
  - 1. Diborides (ZrB<sub>2</sub>, TiB<sub>2</sub>, CrB<sub>2</sub>, MnB<sub>2</sub> and YB<sub>2</sub>).
  - 2.  $Al_2O_3$ -GeO<sub>2</sub> system.
  - 3. Pb  $(Mg_{1/3}Nb_{2/3})0_3$  and related perovskites.
- III. New low thermal expansion magnetic materials:
  - 1. Mono-phasic magnetic [NZP]-compositions.
  - 2. Diphasic composite approach to produce low  $\alpha$  ceramics with ferrimagnetic properties.
- IV. Development of di/multiphasic micro-composite ceramics for low  $\alpha$  applications: The systems studied include [NZP] or [CZP]+Nb<sub>2</sub>0<sub>5</sub>, ZrSiO<sub>4</sub>, GdPO<sub>4</sub>, Mg<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>, Zn<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>, MgO and ZnO.

V. New methods for measuring ultra-low thermal expansion: In this area, with the aid of laser technology, several new techniques for the measurements of very low  $\alpha$  of ceramic materials were developed.

During the three years (July 1983-June 1986) of the contract period of the project entitled, "Ultra-low Thermal Expansion Ceramics," most of the research work stated in the original proposal has been successfully completed. In this final report, a summarized version of the past three annual reports in a cumulative form is being presented.

#### I. Synthesis, Crystal Chemistry, and Thermal Expansion of [NZP]-Family

In this area, several aspects of research include synthesis, characterization, thermal expansion, single crystal growth, crystal chemistry, development of a glass-ceramic, dielectric properties, etc. It was established (1) that the sol-gel process is superior to the solid state reaction method (powder mixing) for synthesizing most of the [NZP]-family members. Further, the traditional sol-gel technique was modified by introducing a seeding step which has improved considerably the sinterability and the microstructure of the material. Single crystals of  $CaZr_4P_6O_{24}$  and  $NaZr_2P_3O_{12}$  were grown by a flux technique, and measurements were made on the crystals supplied by Perrotta (ALCOA); (2) by using facilities at Geophysical Laboratories, Washington, DC. A systematic investigation of the thermal expansion of alkaline [NZP]  $(M^1Zr_2P_3O_{12}$  where  $M^1$  = Li, Na, K, Rb, Cs) and alkaline earth [NZP] ( $M^{11} = Zr_4P_6O_{24}$  where  $M^{11} = Mg$ , Ca, Sr. Ba) was completed (3.4). A structural model (5) to interpret the thermal expansion data in these systems and based on structural parameters of [NZP], was developed; this model can explain the anisotropic thermal expansion of other [NZP]-family members.

It has been shown that  $CaZr_4P_6O_{24}$  (CaZP) and  $SrZr_4P_6O_{24}$  (SrZP) which are structurally isomorphous with  $NaZr_2P_3O_{12}$  [NZP] have low bulk thermal expansion, in fact, CaZP exhibits low negative CTE (coefficient of thermal expansion,  $\alpha$ )

and SrZP low positive  $\alpha$  between 25° and 500°C. However, a new discovery about these materials is that the anisotropies in their axial thermal expansions are reversed, i.e., in CaZP c-axis expands and a-axis contracts while on the other hand the SrZP c-axis contract and a-axis expands. This has been demonstrated in Figure 1 also, in which the lattice parameters 'a' and 'c' have been plotted against temperature. The lattice parameters were determined using powdered material and high temperature X-ray diffractometry. In order to control the thermal expansion properties of a <u>crystalline</u> ceramic material, it is essential that the material exhibits <u>both</u> a low net coefficient of thermal expansion, and a low <u>anisotropy</u> of thermal expansion. The latter is very important for controlling thermal shock in sintered ceramics. We have reduced this to practice and show (see Figure 2) that the <u>anisotropy</u> of  $\alpha$  at Ca<sub>0.5</sub>Sr<sub>0.5</sub>Zr<sub>4</sub>P<sub>6</sub>O<sub>24</sub> is in fact very close to zero. This is an important discovery, a U.S. patent has been filed for this work.

Several compositions in the [NZP]-family were considered for the development of a glass-ceramic and finally,  $NaGe_2P_3O_{12}$  because of its low melting temperature ( $\sim$  1125°C), was found suitable for this purpose (6). Dielectric measurements made on CZP+ MgO, ZnO system revealed that, in general, [NZP] materials have low dielectric constants with relatively low loss.

The crystal chemistry of [NZP] or [CTP]-family (7) was investigated by making numerous substitutions at various lattice sites of [NZP]-structure and also with the aid of published data on this subject. The [NZP]-structural family has an extraordinary range of discrete compositions and crystalline solutions, these compositions being classified according to their crystal chemical substitution schemes.

The effect of polycrystal constraints on the thermal expansion of a material was studied using a powdered and sintered sample of  $NaZr_2P_3O_{12}$  (8).

#### II. Exploration of New Families of Low Expansion Materials

In this area, three different systems, namely diborides, study the anisotropy  $Al_2O_3$ -GeO<sub>2</sub> and  $Pb(Mg_1/3Nb_2/3)O_3$ --perovskite families, were studies in order to look for any new low or near zero expansion compositions. Several diborides, namely  $ZrB_2$ ,  $TiB_2$ ,  $CrB_2$ ,  $MnB_2$  and  $YB_2$ , were chosen, and their axial thermal expansions (9) were measured by high temperature X-ray diffractometry to study the anisotropy in thermal expansion. Several compositions in  $Al_2O_3$ -GeO<sub>2</sub> were synthesized by a powder mixing technique, and their thermal expansions were determined using a Harrop Dilatometric Analyzer. In the perovskite family, synthesis and thermal expansion property measurements of several new compositions such as PMN, modified PMN (10), PZT, and their solid solutions were done. Most of the compositions exhibited large thermal expansions; however, some PMN and PZN based compounds demonstrated very low  $\alpha$  at low temperatures.

#### III. New Low Thermal Expansion Magnetic Materials

Two approaches were adopted to produce new magnetic materials with low thermal expansion characteristics: (i) development of mono-phasic [NZP] compositions by incorporating rare-earth ions  $(Y^{3+}, Gd^{3+})$ , Fe<sup>3+</sup> and Cr<sup>3+</sup> at Na, Zr, or P sites. Only Cr<sup>3+</sup> could be substituted for Zr with excess Na or Ca (11) for charge compensation; (ii) using the diphasic approach in which a suitable [NZP]-compound with negative  $\alpha$  is mixed with a thermodynamically compatible magnetic material, and a ceramic is fabricated by suitable heat treatment. Development of such a ceramic with 85% Na<sub>4</sub>Zr<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> + 15% YIG (Yttrium Iron Garnet) which displays very low thermal expansion up to 150°C (12) has been successful.

IV. Development of Di/Multiphasic Micro-composite Ceramics for Low α Applications

The diphasic micro-composite approach provides an avenue for more firmly

controlled "tailoring" of properties of the materials. The desirable properties of two different materials can be brought together. For this purpose, the specific systems studied include [NZP] or [CZP]+Nb<sub>2</sub>0<sub>3</sub>, GdP0<sub>4</sub>, ZrSi0<sub>4</sub>, Mg<sub>3</sub> (P0<sub>4</sub>)<sub>2</sub>, Zn<sub>3</sub>(P0<sub>4</sub>)<sub>2</sub>, Mg0 and Zn0. Several compositions exhibited a near zero expansion profile over a wide temperature range (13).

#### V. New Methods for Measuring Ultra-low Thermal Expansion

Since the development of a new class of materials with ultra-low thermal expansion, the conventional push-rod dilatometers have become obsolete due to their limitations in measuring precisely, and with high accuracy, very low thermal expansion behavior. To resolve this problem, several new measuring techniques (14-17) utilizing laser spectroscopy have been developed in the past three years. These techniques are highly sensitive, precise, and furnish the data with high accuracy. They are capable of measuring ultra-low thermal expansion of ceramic materials.

#### REFERENCES

- "Improvement of Crystallinity of KZr<sub>2</sub>P<sub>3</sub>O<sub>12</sub> (KZP) by Sol-Gel Synthesis," G. Lenain, H.A. McKinstry, and D.K. Agrawal, J. Am. Cer. Soc., <u>68</u>(9), C-224 (1985).
- 2. "High Temperature Crystal Chemistry of Sodium Zirconium Phosphate [NZP]," L.W. Finger, R.M. Hazen, D.K. Agrawal, H.A. McKinstry, and A.J. Perrotta, J. Mat. Res. (submitted).
- 3. "Low Thermal Expansion Alkali Zirconium Phosphates," G. Lenain, H.A. McKinstry, S. Limaye, and A. Woodword, Mat. Res. Bull., 19 (10), 1451-1456 (1984).
- 4. "Synthesis and Thermal Expansion of MZr<sub>4</sub>P<sub>6</sub>O<sub>24</sub> (M = Mg, Ca, Sr, and Ba)," S.Y. Limaye, D.K. Agrawal, and H.A. McKinstry, J. Am. Cer. Soc. (in process).
- 5. "Crystal Structural Models of Thermal Expansion in [NZP]," G. Lenain, H.A. McKinstry, J. Alamo, and D.K. Agrawal, J. Mats. Sci. (in press).
- 6. "Synthesis and Thermal Expansion of NaGe<sub>2</sub>P<sub>3</sub>O<sub>12</sub>," D.K. Agrawal, K.R. Kanak, and H.A. McKinstry, Mat. Res. Bull. (submitted).
- 7. "Crystal Chemistry of the NaZr<sub>2</sub>P<sub>3</sub>O<sub>12</sub>, [NZP] or [CTP] Structural Family," J. Alamo and R. Roy, J. Mat. Sci.,  $\underline{21}$ , 446-450 (1986).
- 8. "Study of Lattice Parameters of Sintered Ceramics and Powdered Sample of Sodium Zirconium Phosphate," J.F. Cloer, H.A. McKinstry, D.K. Agrawal, and S.Y. Limaye, 88th Annual Meeting of Am. Ceram. Soc., Chicago, May (1984).
- 9. "Relationship Between Thermal Expansion and Crystal Chemical Parameters in Diborides," H.A. McKinstry, R.V. Sara, and K.E. Spear, Adv. in X-ray Anal. (1986) (in process).
- 10. "Thermal Expansion of B-Site Modification of Lead Magnesium Niobate," D.K. Agrawal and R. Roy, 88th Annual Meeting of Am. Ceram. Soc., Chicago, May (1986).

- 11. "Synthesis and Thermal Expansion of  $Na_{3-x}Ca_{x}/2M_{2}P_{3}O_{12}$  (M = Fe, Cr)," D.K. Agrawal and R. Roy, J. Mat. Sci. Let., 5, 139-140 (1986).
- 12. "New Low Expansion Magnetic Materials a Composite Approach," D.K. Agrawal and R. Roy, Proc. 21st University Conference on Ceramic Science, Penn State University, July (1985) (in press).
- 13. "Composite Route to 'Zero' Expansion Ceramics," D.K. Agrawal and R. Roy, J. Mats. Sci., 29, 4617-4623 (1985).
- 14. "A Laser Speckle Method for Measuring Thermal Expansion," C.S. Vikram and H.A. McKinstry, Optics and Lasers in Engineering  $\underline{6}(2)$ , 91-100 (1985).
- 15. "A Simple Laser Speckle Dilatometer for Thermal Expansion Measurements," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, Mats. Let. 3(12), 482-484 (1985).

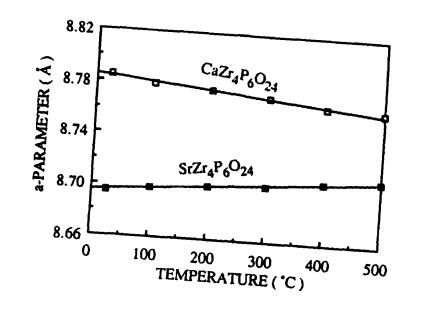
- 16. "Thermal Expansion Measurements by Photodetector Based Laser Beam Position Change Determination," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, Mat. Res. Soc. Symp. Vol. 50, 1986 (in press).
- 17. "Application of Dial-gage and Laser Speckles for Ultra-low Thermal Expansion Measurements using Push-Rod Dilatometers," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, J. Mat. Res. (submitted).

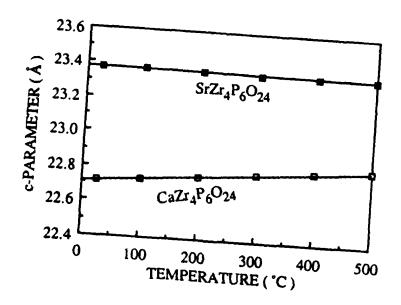
#### LIST OF PUBLICATIONS

- 1. "Crystal Chemistry of the NaZr<sub>2</sub>P<sub>3</sub>O<sub>12</sub>, [NZP] or [CTP] Structural Family," J. Alamo and R. Roy, J. Mat. Sci., <u>21</u>, 444-450 (1986).
- 2. "Low Thermal Expansion Alkali Zirconium Phosphates," G. Lenain, H.A. McKinstry, S. Limaye, and A. Woodword, Mat. Res. Bull, 19(10), 1451-1456 (1984).
- 3. "Crystal Structural Models of Thermal Expansion in NZP," G. Lenain, H.A. McKinstry, J. Alamo, and D.K. Agrawal, J. Mats. Sci. (in press).
- 4. "A Laser Speckle Method for Measuring Thermal Expansion," C.S. Vikram and H.A. McKinstry, Optics and Lasers in Engineering, 6(2), 91-100 (1985).
- 5. "Composite Route to 'Zero' Expansion Ceramics," D.K. Agrawal and R. Roy, J. Mats. Sci., 20, 4617-4623 (1985).
- 6. "Successful Design of New Very Low Thermal Expansion Ceramics," R. Roy and D.K. Agrawal, Mat. Res. Soc. Symp. Proc., Vol, 40, Eds. E.A. Giess, King-Ning Tu, and E.A. Uhlman, 83-88 (1985), Mats. Res. Soc. Publication.
- 7. "Improvement of Crystallinity of KZr<sub>2</sub>P<sub>3</sub>O<sub>12</sub> (KZP) by Sol-Gel Synthesis," G. Lenain, H.A. McKinstry, and D.K. Agrawal, J. Am. Cer. Soc., <u>68</u>(9), C-224 (1985).
- 8. "Low Thermal Expansion of Alkali Zirconium Phosphates Using a Microcomputer Automated Diffractometer," G.E. Lenain, H.A. McKinstry, and S.Y. Limaye, Adv. in X-ray Anal., 28, Eds. Barrett, Predecki, and Leyden, pp. 345-352 (1985).
- 9. "Synthesis and Thermal Expansion of Na<sub>3-x</sub>Ca<sub>x</sub>/<sub>2</sub>M<sub>2</sub>P<sub>3</sub>O<sub>12</sub> (M = Fe, Cr)," D.K. Agrawal and R. Roy, J. Mat. Sci. Let., 5, 139-140 (1986).
- 10. "A Simple Laser Speckle Dilatometer for Thermal Expansion Measurements," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, Mats. Let., 3(12), 482-484 (1985).
- 11. "Thermal Expansion of  $NH_4Zr_2(PO_4)_3$ ," S. Komarneni, G.E. Lenain, and R. Roy, J. Mat. Sci. Let., 5, 1-3 (1986).
- 12. "Laser Speckles for Fine Measurements," In Computer Aided Testing and Model Analysis, C.S. Vikram, pp. 31-33, Proc. 1984 SEM Fall Conf., Milwaukee, WI, Pub. by Soc. for Experimental Mechanics, CT (1984).
- 13. "Synthesis and Thermal Expansion of MZr<sub>4</sub>P<sub>6</sub>O<sub>24</sub> (M = Mg, Ca, Sr, and Ba)," S.Y. Limaye, D.K. Agrawal, and H.A. McKinstry, J. Am. Cer. Soc. (in process).
- 14. "New Low Expansion Magnetic Materials a Composite Approach," D.K. Agrawal and R. Roy, in Tailoring Multi-phasic Ceramics, pp. 381-384, Plenum Pub. Corp. (1986) (in press).
- 15. "Study of Lattice Parameters of Sintered Ceramic and Powdered Samples of Sodium Zirconium Phosphate," J.F. Cloer, H.A. McKinstry, D.K. Agrawal, and S.Y. Limaye, Mat. Res. Bull. (in process).

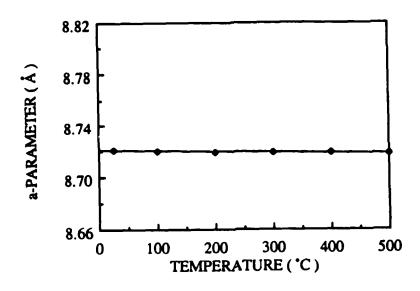
- 16. "Synthesis and Thermal Expansion of NaGe<sub>2</sub>P<sub>3</sub>O<sub>12</sub>," D.K. Agrawal, K. Kanak, and H.A. McKinstry, J. Mat. Sci. Let. (submitted).
- 17. "High Temperature Crystal Chemistry of Sodium Zirconium Phosphate [NZP]," L.W. Finger, R.M. Hazen, D.K. Agrawal, H.A. McKinstry, and A.J. Perrotta, J. Mat. Sci. (submitted).
- 18. "Thermal Expansion Measurement by Photodetector Based Laser Beam Position Change Determination," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, Mat. Res. Soc. Symp. Proc., Vol. 50 (in press).
- 19. "Application of Dial-gage and Laser Speckles for Ultra-low Thermal Expansion Measurements using Push-Rod Dilatometers," C.S. Vikram, D.K. Agrawal, R. Roy, and H.A. McKinstry, J. Mat. Res. (submitted).
- 20. "Relationship Between Thermal Expansion and Crystal Chemical Paramaters in Diborides," H.A. McKinstry, R.V. Sara, and K.E. Spear, Adv. in X-ray Anal. (1986) (in press).
- 21. "Thermal Expansion of B-Site Modifications of Lead Magnesium Niobate," D.K. Agrawal, R. Roy, and H.A. McKinstry, J. Mat. Res. (in process).

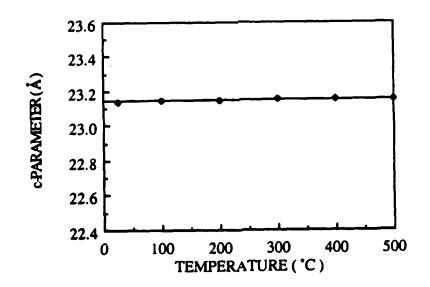
# AXIAL THERMAL EXPANSION $CaZr_4P_6O_{24}$ AND $SrZr_4P_6O_{24}$





## AXIAL THERMAL EXPANSION Ca<sub>0.5</sub>Sr<sub>0.5</sub>Zr<sub>4</sub>P<sub>6</sub>O<sub>24</sub>





DTZ

9 49 3